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596 Chapter 19 Acids and Bases PRACTICE PROBLEMS 1. Write a balanced formula equation for the reaction that occurs between each of the following pairs of reactants. a. magnesium and nitric acid b. aluminum and sulfuric acid c. calcium carbonate and hydrobromic acid d. potassium hydrogen carbonate and hydrochloric acid

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of acid or base are added. A buffer system should contain considerably more acid than base. Specific buffer systems should be chosen based on the pH that must be maintained. 114 Study Guide for Content Mastery Chemistry: Matter and Change Chapter 19

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Chapter 19 Acids, Bases, and Salts Flashcard What are the properties of acids and bases? acids taste sour, will change the color of an acid-base indicator, and can be strong or weak electrolytes in aqueous solution.

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Study Guide: Acids and Bases 1) List at least three characteristic properties of acids and three of bases. ACIDS pH less than 7 Have a sour taste Change the color of many indicators Are corrosive (react with metals) Neutralize bases Conduct an electric current BASES

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