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13. Why might a construction company build a high-rise building with steel alloy beams instead of pure . iron beams? _____

14. Water and ethanol are miscible in all proportions, whereas oil and water are immiscible. Explain this . statement.

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Factor #1 Concentration If more molecules of "A" are added to the system at equilibrium, then there will be more collisions between "A" molecules and the remaining "B" molecules, producing more the products, "C" and "D," causing the equilibrium to shift to the right.

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24—1 Review and Reinforcement (continued) 13. There are 3.29 g of iodine-126 remaining in a sample originally containing 26.3 g of iodine-126. The half-life of iodine-126 is 13 days.

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10—2 Review and Reinforcement (continued) 9. 0.0621 mol SS = (Q \$ gK2S 10. 65.8gMg= l. atoms Mg Solve each of the following

problems as directed. Show all your work. 11. Calculate the number of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NO₂ gas.

10-2 Review and Reinforcements - Front Page

Name _____ I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles of magnesium are required to react with 2.0 mol of hydrochloric acid?

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Chemistry: Chapter 15 Concentration and Nature of Solutions

13. luz visible 14. fotón Reinforcement (page 27) Section 1 1. Both are caused by something vibrating, and both transfer energy. Sound waves are compressional waves. they must have a medium to transfer energy. Electromagnetic waves are transverse waves. They can travel without a medium. 2. Electric charges vibrate or oscillate. 3.

Name _____ Date _____ Class 1 Reinforcement What are electromagnetic ...

3—3 Review and Reinforcement Modern Atomic Theory Complete the following table. 1. 2. 3. If the statement is true, write “true.” If it is false, change the underlined word or words to make the statement true. Write your answer on the line provided. 4. Moseley discovered that all atoms of an element have the same number of neutrons in ...

Name _____ Date _____ Class 3—3 Review and Reinforcement

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erty Gas Condensed States of Matter. Fill in the 14-1 Review and Reinforcement ...

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